

Flexible synthesis of montanine-like alkaloids: revisiting the structure of montabuphine†

Yifu Guan, Hongbin Zhang,* Chengxue Pan, Jia Wang, Rong Huang and Qilin Li

Received 21st February 2012, Accepted 23rd March 2012

DOI: 10.1039/c2ob25374g

An efficient and stereocontrolled synthetic strategy towards the synthesis of montanine-like alkaloids was developed. Our results suggest that the structure elucidation for natural montabuphine needs further elaboration.

In the *Amaryllidaceae* alkaloids, there is a small family of natural products, namely the montanine-type alkaloids, bearing a unique pentacyclic 5,11-methanomorphanthridine skeleton (Fig. 1).¹ These alkaloids encompass anxiolytic, anti-depressive, anti-convulsive and weak hypotensive activities.² The unique structural features and important biological activities associated with these alkaloids have attracted considerable synthetic efforts.³ Recently, Banwell reported an elegant strategy towards the synthesis of montanine-type alkaloids.^{4a} Although they successfully synthesized the assigned structure for montabuphine^{4b} (5, Fig. 1), unfortunately the physical and spectral data of this compound do not match those recorded for the natural product. Their result established that the structure of montabuphine needs to be revised.⁵ Based on Banwell's work, we deduced that the two structures, with a *trans* spatial arrangement for the C4a-H and the aromatic ring, as highlighted in Fig. 1 (6 and 7) might be the alternative possibility for montabuphine. Although most of the montanine-type alkaloids have been synthesized, few efforts have been directed towards the synthesis of montanine-like diastereoisomers with a C4a-H orientated *trans* to the aromatic ring. Herein, we report the stereocontrolled synthesis of these structures (6 and 7).

Our flexible approach towards the synthesis of the 5,11-methanomorphanthridine alkaloids is outlined in Scheme 1. Developing a regioselective and stereocontrolled palladium-catalyzed cross coupling of vinyl epoxides (11 and 12) with organoboronic acid 13 is the key issue⁶ for the synthesis of montabuphine structure (6 or 7) proposed in Fig. 1.

Key Laboratory of Medicinal Chemistry for Natural Resource, Ministry of Education, School of Chemical Science and Technology, Yunnan University, Kunming, Yunnan 650091, P. R. China. E-mail: zhanghb@ynu.edu.cn, zhang_hongbin@hotmail.com

†Electronic supplementary information (ESI) available: ¹H NMR, ¹³C NMR spectra of new products and experimental details. CCDC 867573 (11), 867574 (12), 867575 (14b), 871431 (14c) and 867576 (15b). For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c1cc11949d

Our research was initiated with preparation of tosylamide 10. Treatment of 2-(4-hydroxyphenyl)ethylamine (8, Scheme 2) with *p*-toluenesulfonyl chloride afforded 8a, which, after oxidative dearomatization with iodobenzene diacetate (IBD)⁷ and subsequent aza Michael addition,⁸ afforded compound 9a in 48% yield in 3 steps. Dehydration of 9a with phosphorus oxychloride in pyridine provided enone 10 in 69% yield.

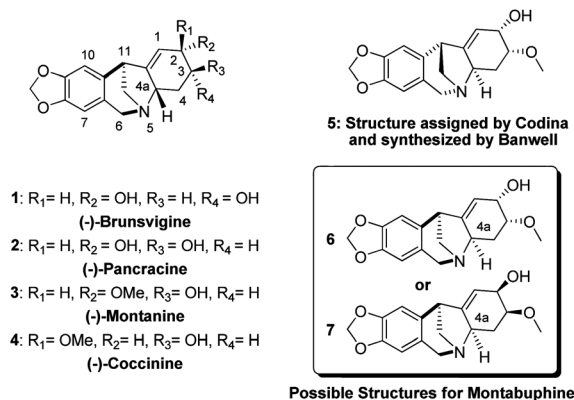
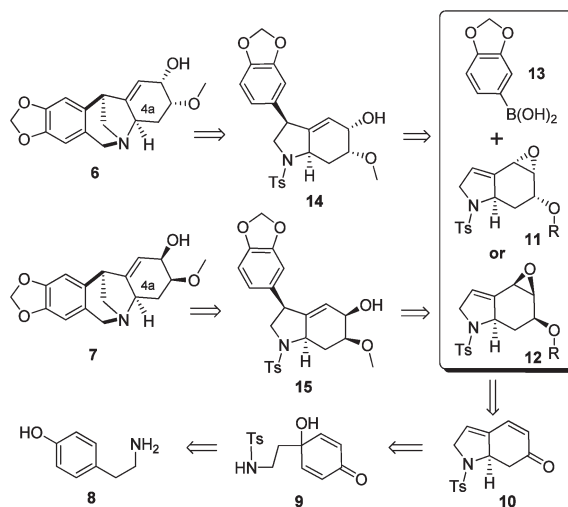
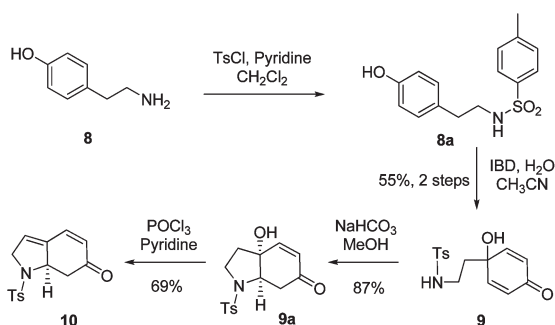


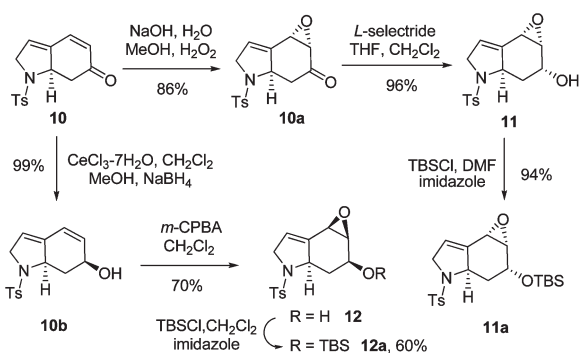
Fig. 1 Speculation for the structure of montabuphine.



Scheme 1 Retrosynthetic analysis of compound 6 and 7.



Scheme 2 Synthesis of enone 10.

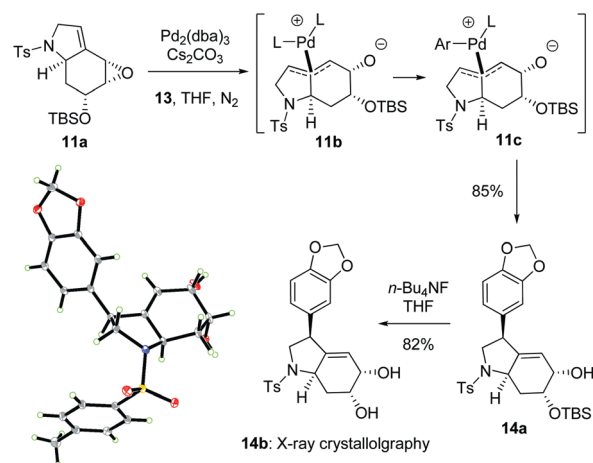


Scheme 3 Stereoselective synthesis of epoxides 11 and 12.

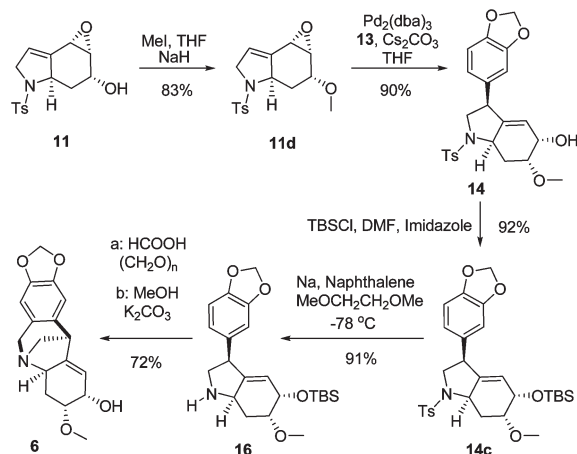
Having obtained enone 10, we next set our goal as the stereocontrolled synthesis of epoxides 11 and 12 (Scheme 3). Treatment of compound 10 with hydrogen peroxide in methanol resulted in ketone 10a stereoselectively. After reduction with *L*-selectride, epoxide 11 was isolated in 96% yield. Protection with *tert*-butyl dimethyl silyl chloride (TBSCl) afforded compound 11a in 94% yield. Reduction of enone 10 under Luche condition followed by epoxidation of the resulting alcohol (10b) with *m*-chloroperbenzoic acid afforded epoxide 12 in 70% yield in two steps. The relative configurations for epoxides 11 and 12 were established by NMR and also confirmed by X-ray crystallography.⁹

Next we came to the key cross-coupling reaction of vinyl epoxide 11a with organoboronic acid 13. Utilization of Szabó's reaction condition, [Pd₂(dba)₃] as catalyst in THF–H₂O (10 : 1) in the presence of Cs₂CO₃, unfortunately gave only small amount (<3%) of desired coupling product.⁶ After some experimentation, the optimal condition was found, treatment of compound 11a, catalyst [Pd₂(dba)₃] and organoboronic acid 13 in the presence of Cs₂CO₃ in anhydrous THF at room temperature afforded the desired product 14a in 85% yield. This reaction was presumed to form initially an (η^3 -allyl)-palladium(II) complex by opening of the epoxide ring.⁶ After transmetalation of organoboronic acid, (Scheme 4, 11b to 11c) the aromatic group was added regioselectively and stereoselectively to the pyrrolidine ring. The relative stereochemistry was unambiguously established by X-ray crystallographic analysis of compound 14b.

O-Methylation of epoxide 11 with methyl iodide in the presence of sodium hydride afforded ether 11d (83%). The palladium(0) complex catalyzed cross coupling of epoxide 11d with



Scheme 4 Stereoselective synthesis of the key intermediate.

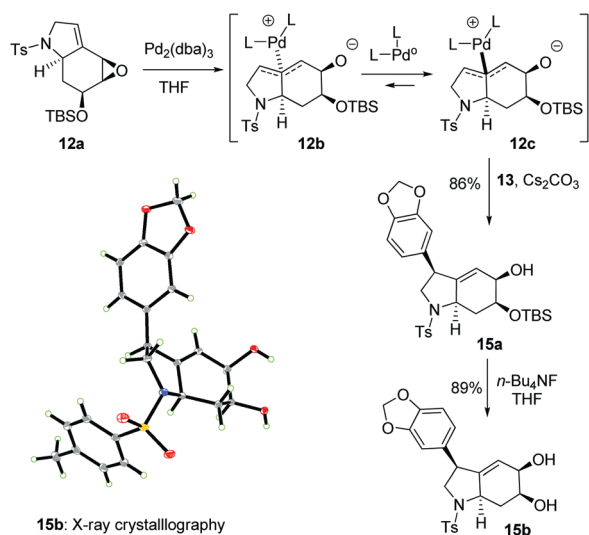
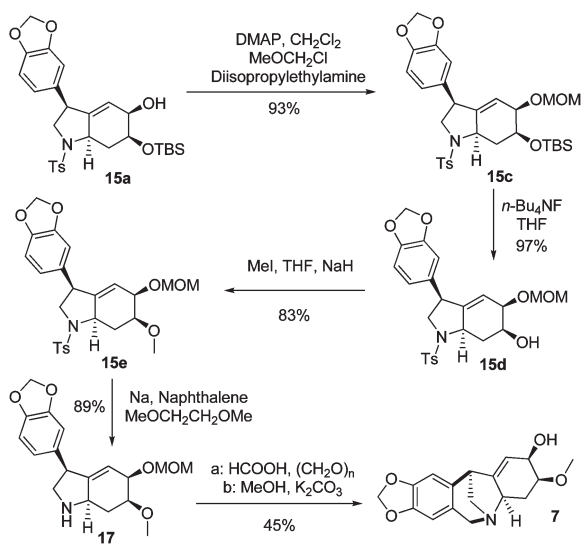


Scheme 5 Synthesis of the deduced structure, compound 6.

organoboronic acid 13 gave the key intermediate (14) in 90% yield. Protection of 14 with TBSCl afforded 14c and the relative stereochemistry of 14c was confirmed by X-ray crystallographic analysis. After detosylation of 14c with sodium naphthalide,^{3j} amine 16 was obtained in 83% yield in two steps (Scheme 5). Finally, treatment of amine 16 with formaldehyde in the presence of formic acid^{4a} resulted in the desired product (6) in 72% yield.

To our disappointment, the spectroscopic data of compound 6 did not match these reported in the literature^{4b} (see Table 1 in the ESI†). We next carried out the synthesis of structure 7. To our delight, the palladium catalyzed reaction of vinyl epoxide 12a with organoboronic acid 13 yielded the desired diastereoisomer (15a) as the sole product (Scheme 5, relative stereochemistry confirmed by X-ray crystallography).

This result could be explained by the mechanism of nucleophilic palladium(0) displacement of the (π -allyl)-palladium intermediate (12b to 12c in Scheme 6) proposed by Bäckvall and Granberg.¹⁰ Next, compound 15a was converted to its methyl ether (15e) in three steps (45% overall yield) by manipulation of the C2 and C3 hydroxyl protecting groups. Detosylation of compound 15e with sodium in naphthalene afforded amine 17 in 89% yield. Finally, utilization of Banwell's procedure furnished

Scheme 6 Synthesis of intermediate **15a**.Scheme 7 Synthesis of the deduced structure, compound **7**.

the compound **7** in 45% yield. In comparison with data recorded in the literature,^{4b} the ¹H and ¹³C NMR spectra of compound **7** were unfortunately found to be different from that reported for montabuphine (Scheme 7).

In conclusion, we have developed an efficient and flexible strategy for the synthesis of montanine-like compounds. Although our results did not establish the true structure for montabuphine, this new approach, features a stereocontrolled palladium-catalyzed cross coupling of vinyl epoxides with organoboronic acids, provided a versatile route towards the synthesis of medicinally interesting montanine-type diastereoisomers. We are currently working on the issue of relative configuration for C2 and C3 in the hope that a true structure could be assigned to natural montabuphine in the near future.

Acknowledgements

This work was supported by grants from Natural Science Foundation of China (20925205, 20832005), National Basic Research Program of China (973 Program 2009CB522300) and Yunnan Provincial Science & Technology Department (2010GA014).

Notes and references

- For reviews, see: (a) S. F. Martin, *The Amaryllidaceae Alkaloids*, in *The Alkaloids*, ed. A. Brossi, Academic Press, San Diego, 1987, vol. 30, p. 252; (b) O. Hoshino, *The Amaryllidaceae Alkaloids*, in *The Alkaloids*, ed. G. A. Cordell, Academic Press, San Diego, 1998, vol. 51, p. 323. For a recent review, see: (c) Z. Jin, *Nat. Prod. Rep.*, 2011, **28**, 1126 and references cited therein.
- (a) F. Viladomat, J. Bastida, C. Codina, W. E. Campbell and S. Mathee, *Phytochemistry*, 1995, **40**, 307; (b) J. Labraña, A. K. Machocho, V. Kricsfalussy, R. Brun, C. Codina, F. Viladomat and J. Bastida, *Phytochemistry*, 2002, **60**, 847; (c) A. F. Schürmann da Silva, J. P. de Andrade, L. R. M. Bevilacqua, M. M. de Souza, I. Izquierdo, A. T. Henriques and J. Á. S. Zuanazzi, *Pharmacol., Biochem. Behav.*, 2006, **85**, 148.
- (a) M. Ishizaki, O. Hoshino and Y. Iitakab, *Tetrahedron Lett.*, 1991, **32**, 7079; (b) L. E. Overman and J. Shim, *J. Org. Chem.*, 1991, **56**, 5005; (c) M. Ishizaki, O. Hoshino and Y. Iitaka, *J. Org. Chem.*, 1992, **57**, 7285; (d) M. Ishizaki, K.-I. Kurihara, E. Tanazawa and O. Hoshino, *J. Chem. Soc., Perkin Trans. 1*, 1993, 101; (e) L. E. Overman and J. Shim, *J. Org. Chem.*, 1993, **58**, 4662; (f) J. Jin and S. M. Weinreb, *J. Am. Chem. Soc.*, 1997, **119**, 5773; (g) W. H. Pearson and B. W. Lian, *Angew. Chem., Int. Ed.*, 1998, **37**, 1724; (h) M. Ikeda, M. Hamada, T. Yamashita, K. Matsui, T. Sato and H. Ishibashi, *J. Chem. Soc., Perkin Trans. 1*, 1999, 1949; (i) C.-K. Sha, A.-W. Hong and C.-M. Huang, *Org. Lett.*, 2001, **3**, 2177; (j) M. G. Banwell, A. J. Edwards, K. A. Jolliffe and M. Kemmler, *J. Chem. Soc., Perkin Trans. 1*, 2001, 1345; (k) G. Pandey, P. Banerjee, R. Kumar and V. G. Puranik, *Org. Lett.*, 2005, **7**, 3713; (l) M. G. Banwell, O. J. Kokas and A. C. Willis, *Org. Lett.*, 2007, **9**, 3503; (m) O. J. Kokas, M. G. Banwell and A. C. Willis, *Tetrahedron*, 2008, **64**, 6444; (n) A.-W. Hong, T.-H. Cheng, V. Raghukumar and C.-K. Sha, *J. Org. Chem.*, 2008, **73**, 7580; (o) M. Anada, M. Tanaka, N. Shimada, H. Nambu, M. Yamawaki and S. Hashimoto, *Tetrahedron*, 2009, **65**, 3069; (p) S. V. Pansare, R. Lingampally and R. L. Kirby, *Org. Lett.*, 2010, **12**, 556; (q) G. Pandey, R. Kumar, P. Banerjee and V. G. Puranik, *Eur. J. Org. Chem.*, 2011, 4571.
- Synthesis of assigned structure for montabuphine, see: (a) M. Matveenko, M. G. Banwell and A. C. Willis, *Org. Lett.*, 2008, **10**, 4693. Montabuphine isolation, see: (b) F. Viladomat, J. Bastida, C. Codina, W. Campbell and S. Mathee, *Phytochemistry*, 1995, **40**, 307.
- There is an uncertainty about the C4a stereochemistry. In the original structure elucidation paper, Codina did not define the stereochemistry at C4a. Banwell synthesized the structure with the C4a hydrogen orientated *cis* to the aromatic ring. While in Pandey's paper, they draw the structure of montabuphine with the C4a hydrogen orientated *trans* to the aromatic moiety; see ref. 3q.
- Recently Szabó published the first carbon-carbon cross coupling of vinyl epoxides with organoboronic acids catalyzed by palladium pincer complex. Only one example involving cyclic vinyl epoxide was reported, however with poor regioselectivity (2:1). See: J. Kjellgren, J. Aydin, O. A. Wallner, I. V. Saltanova and K. J. Szabó, *Chem.-Eur. J.*, 2005, **11**, 5260 and references cited therein.
- F.-X. Felpin, *Tetrahedron Lett.*, 2007, **48**, 409.
- Recently, You's group reported an asymmetric synthesis of intermediate **9a**, see: Q. Gu and S.-L. You, *Chem. Sci.*, 2011, **2**, 1519.
- CCDC 867573 (for **11**), CCDC 867574 (for **12**), CCDC 867575 (for **14b**), CCDC 871431 (for **14c**) and CCDC 867576 (for **15b**) contain the supplementary crystallographic data for this paper.
- K. L. Granberg and J.-E. Bäckvall, *J. Am. Chem. Soc.*, 1992, **114**, 6858.